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# **Short Communication**

# High-performance liquid chromatographic determination of the antibiotic cortalcerone

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#### ABSTRACT

Cortalcerone (2-hydroxy-6H-3-pyrone-2-carboxaldehyde hydrate) is an antibacterial antibiotic produced by a number of basidiomycetous fungi. Its biosynthesis is connected with C-2 oxidation of Dglucose. A method for the reliable determination of this antibiotic is required for the study of the mechanism of its biosynthesis and for the determination of its pharmacological properties. The high-performance liquid chromatographic determination of cortalcerone in the presence of D-glucose, D-fructose, D-*arabino*-2-hexosulose, D-gluconic acid and D-*arabino*-2-hexulosonic acid on an amino-phase column with 70% aqueous acetonitrile (pH 5.00) and UV detection at 195 nm is reported. A linear relationship between response and cortalcerone concentration was found in the range 0.25–10.00 mg/ml. The detection limit is  $6.5 \mu g/ml$ .

#### INTRODUCTION

Cortalcerone, an antibacterial antibiotic produced by a number of basidiomycetous fungi [1,2], is a  $\beta$ -pyrone compound unusual in nature, 2-hydroxy-6H-3-pyrone-2-carboxaldehyde hydrate (Fig. 1). Its biosynthesis starts from D-glucose and consists of two enzymatic steps; in the first step; D-glucose is oxidized by the action of the enzyme pyranose 2-oxidase into D-*arabino*-2-hexosulose (D-glucosone) [3], which is then dehydrated to cortalcerone by a new enzyme, pyranosone dehydratase [4] via putative unstable intermediates that have not yet been identified.

Fig. 1. Structure of cortalcerone.

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A method for the reliable determination of cortalcerone is required for the study of the mechanism of its biosynthesis. High-performance liquid chromatography (HPLC) on silica [5], ion-exchange [6], reversed-phase [7] or amino-phase [8] columns has often been used for the determination of sugars and/or their derivatives in natural samples. The separation of sugars on various stationary phases has been reviewed and discussed in detail [9,10]. For cortalcerone, only a qualitative thin-layer chromatographic assay is available [2].

A simple HPLC method has been devised for the determination of cortal cerone in the presence of D-glucose, D-fructose, D-*arabino*-2-hexosulose, D-gluconic acid and D-*arabino*-2-hexulosonic acid using an amino-phase column and UV photometric detection. The method was used to examine the time course of cortal cerone production by the washed intact mycelium of the basidiomycete *Phanerochaete chrysosporium* during incubation with a solution of pure D-glucose.

#### **EXPERIMENTAL**

# Chemicals

Cortalcerone was isolated [11] from submerged cultures of the basidiomycete *Phanerochaete chrysosporium* (CCBAS 571, Culture Collection of Basidiomycetes, Institute of Microbiology, Prague (Czechoslovakia) by a procedure described by Baute *et al.* [1]. D-*arabino*-2-hexosulose was synthesized enzymatically [12] from D-glucose. D-Glucose was obtained from Fluka (Buchs, Switzerland), D-fructose was of analytical-reagent grade from Lachema (Brno, Czechoslovakia), sodium D-gluconate was purchased from Koch-Light Labs. (Haverhill, U.K.) and calcium D-*arabino*-2-hexulosonate from Merck (Darmstadt, Germany).

#### Sample preparation

All standards were prepared as aqueous solutions of concentration 10 mg/ml unless noted otherwise. The time course of cortalcerone production was studied as follows: 10 g of washed intact mycelium of the basidiomycete *Phanerochaete chrysosporium* were suspended in 50 ml of a 2% aqueous solution of D-glucose. The suspension was shaken at 20°C for 48 h, than 100- $\mu$ l samples of liquid were taken and, after filtration through an Acrodisc 13 disposable filter unit (Gelman Sciences, Ann Arbor, MI, U.S.A.), 10- $\mu$ l aliquots were chromatographed.

# Column liquid chromatography

An HP 1090 liquid chromatograph with a diode-array detector (Hewlett-Packard, Amstelveen, The Netherlands) was used. For quantitative purposes compounds were detected at 195 nm. The separation was performed on an analytical column (200 mm  $\times$  4 mm I.D.) packed with Separon SGX NH<sub>2</sub>, 7  $\mu$ m (Tessek, Prague, Czechoslovakia). The mobile phase was acetonitrile-water (70:30) of pH 5.0 adjusted by addition of 0.1 *M* phosphate buffer (1 ml of buffer to 1 l of the mobile phase), at a flow-rate of 1.5 ml/min. Quantification was performed by recording the ratio of the peak area to that of calibration standards of cortalcerone injected directly into the HPLC column.

## Precision and reproducibility studies

The precision and linearity were validated by analysing ten calibration standards containing 0-10.00 mg/ml of cortalcerone in triplicate over 48 h. The reproducibility of the peak-area ratios over the calibration range, expressed as relative standard deviation (R.S.D.), ranged from 5.2 to 7.0%. The R.S.D. of the peak-area ratio was determined at each concentration level of the calibration graph. The slope and the correlation coefficient of the calibration graphs were also calculated.

# **RESULTS AND DISCUSSION**

Cortalcerone is a relatively unstable compound that recyclizes spontaneously to 2-furylglyoxal. In the pH range 5.0–7.0 cortalcerone forms an equilibrium mixture with 2-furylglyoxal which contains 85.6% of cortalcerone. These solutions are stable for 2 days under laboratory conditions. The heat lability of cortalcerone precludes the use of ion-exchange columns which are usually operated at elevated temperatures. For this reason and the earlier successful separation of D-*arabino*-2-hexosulose on an amino-bonded column [8], this type of stationary phase was chosen. The formation of Schiff bases of carbonyl compounds with the column packing [13] can be overcome by using a mobile phase buffered on the acidic side.

Dicarbonyl sugars in solution possess a number of tautomeric forms [14,15]. To check possible complications resulting from different chromatographic behaviours of different forms of D-*arabino*-2-hexosulose, we sampled aliquots of solutions prepared in 0.1 M acetate buffers of pH 4.0, 4.5, 5.0, 5.5, 6.0 and 6.5 after equilibration for 2 h. Comparison of the chromatograms showed that negligible changes in the retention time and peak area of D-*arabino*-2-hexosulose occurred. As D-*arabino*-2-hexosulose decomposes in the presence of trace amounts of a base [16], the mobile phase should be buffered on the acidic side.

D-Glucose and D-arabino-2-hexosulose coelute when the mobile phase is not buffered or when phosphate buffer of pH 6.0 is used. Considering the stability of cortalcerone and D-arabino-2-hexosulose, a mobile phase of pH 5.0 was adopted. In the concentration region between 60 and 80% of acetonitrile in the mobile phase, the best resolutions of both D-glucose–D-arabino-2-hexosulose and cortalcerone–2-furyl-



Fig. 2. Chromatogram of (I) 2-furylglyoxal, (II) cortalcerone, (III) D-fructose, (IV) D-glucose, (V) Darabino-2-hexosulose, (VI) D-arabino-2-hexulosonic acid and (VII) D-gluconic acid using UV detection at 195 nm.



Fig. 3. Production curve of cortaleerone by the washed intact mycelium of the basidiomycete *Phanerochaete chrysosporium*.

glyoxal were obtained when 70% of acetonitrile of pH 5.0 was used; reliable measurements were then achieved.

Fig. 2 shows the measurement of cortalcerone using UV detection at 195 nm in the presence of D-glucose, D-fructose, D-*arabino*-2-hexosulose, D-gluconic acid and D-*arabino*-2-hexulosonic acid. These compounds could play a role in the biochemical oxidation of D-glucose in fungi. A linear relationship was found between the peak area at 195 nm and the cortalcerone concentration in the range 0.25–10.00 mg/ml (slope 0.267 absorbance/ $\mu$ g, correlation coefficient 0.9995, R.S.D. 6.3% from five parallel determinations). At a signal-to-noise ratio of 3, the detection limit was found to be 6.5  $\mu$ g/ml.

The method was used to determine the time course of cortalcerone production by the washed intact mycelium of the basidiomycete *Phanerochaete chrysosporium*. The production curve is shown in Fig. 3. The maximum cortalcerone concentration was found after incubation for 26 h.

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